

Educational Session 1	10.30 - 11.30
Short Break	11.30 - 11.40
Educational Session 2	11.40 - 13.00
Lunch	13.00 - 14.00
Opening & Session 1	14.00 - 15.40





« High field » NMR Spectroscopy

Liquid (high resolution) state

« Imaging » MRI



« Low-field » TD NMR





EPR Spectroscopy/Imaging





1000 750 500 250 0

ppm



# Back to basics













## NMR relaxation :



#### NMR relaxation : return to equilibrium



#### NMR relaxation : return to equilibrium



$$\frac{dM_{x,y}^{t}}{dt} = -R_2 M_{x,y}^{t} \qquad \qquad M_{x,y}^{t} = M_{x,y}^{0} e^{-R_2 t} \qquad \qquad R_2 = ($$

ordered

(T<sub>2</sub>)<sup>-1</sup> disordered

coherent

loss of order loss of coherence

incoherent

T<sub>2</sub> relaxation **Entropic process** Mutual exchange of energy (spin-spin) Can be caused by chemical exchange as well!



## Importance of T<sub>1</sub> and T<sub>2</sub> for spectroscopy

# Importance of $T_1$ and $T_2$ for spectroscopy



### Importance of $T_1$ and $T_2$ for spectroscopy

The NMR signal attenuates exponentially with time

$$s(t) \sim M_A \cos \omega_A t(e^{-t/T_2})$$

Small molecules :  $T_2 \,^\sim 0.2$  to 2s

Beyond a time equal to 3 times  $T_2$  88% of the signal has dissipated and mostly noise is measured

The value of  $T_2$  thus determines the time period during which the signal should be monitored



As the value of T<sub>2</sub> shortens, the resonance line will broaden

#### Importance of $T_1$ and $T_2$ for spectroscopy



As the molecular weight increases, the lines become much broader, affecting resolution and information contents

## Importance of $T_1$ and $T_2$ for spectroscopy

Spin echo sequence



Polystyrene with Mr 50.000 and camphor,  $\rm T_2~PS$  ~ 5ms <<  $\rm T_2$  camphor ~0.5s









Collection of <sup>1</sup>H and <sup>13</sup>C signals





#### Structure Analysis & Elucidation

Scalar coupling interaction between chemically nearby  ${}^1\!Hs$ 



<sup>1</sup>H-<sup>1</sup>H COSY : Correlating neighbouring hydrogens

- Individual <sup>1</sup>H's must be within three chemical bonds
- Only <sup>1</sup>H's from the same unit are correlated due to the glycosidic links

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Scalar coupling interaction between directly bonded <sup>1</sup>H and <sup>13</sup>C



<sup>1</sup>H-<sup>13</sup>C HSQC : Single Bond Correlation

- Establishing C-H bonding pairs in the molecule

- No quaternary carbons
- $-CH_3$ ,  $CH_2$ , CH can be independently established (meHSQC)





Scalar coupling interaction between nearby  $^1\mathrm{Hs}$  and  $^{13}\mathrm{Cs}$ 



<sup>1</sup>H-<sup>13</sup>C HMBC : Multiple Bond Correlation

- Quaternary carbons identified
- Allows the immediate vicinity of the carbon skeleton to be inferred
- Connections across glycosidic linkages established









## Structure Analysis & Elucidation

#### The NOESY spectrum as the basis for solution structure determination

The molecular structure, defined at the level of a dense network of inter hydrogen distances, is thus encoded in the 2D NOESY spectrum of a molecule



NMR based structure determination consists in extracting the dense network of distances from the NOESY spectrum and computing the corresponding 3D structure